

# CPC COOPERATIVE PATENT CLASSIFICATION

## C CHEMISTRY; METALLURGY

(NOTES omitted)

### CHEMISTRY

#### C01 INORGANIC CHEMISTRY

(NOTES omitted)

**C01C AMMONIA; CYANOGEN; COMPOUNDS THEREOF** ({metal hydrides, monoborane, diborane or addition complexes thereof [C01B 6/00](#)}; salts of oxyacids of halogens [C01B 11/00](#); peroxides, salts of peroxyacids [C01B 15/00](#); thiosulfates, dithionites, polythionates [C01B 17/64](#); compounds containing selenium or tellurium [C01B 19/00](#); azides [C01B 21/08](#); {compounds other than ammonia or cyanogen, containing nitrogen, non-metals and optionally metals [C01B 21/082](#)}; metal imides or amides [C01B 21/092](#); nitrites [C01B 21/50](#); {compounds of noble gases [C01B 23/0005](#)}; phosphides [C01B 25/08](#); salts of oxyacids of phosphorus [C01B 25/16](#); compounds containing silicon [C01B 33/00](#); compounds containing boron [C01B 35/00](#))

#### WARNING

In this subclass non-limiting references (in the sense of paragraph 39 of the Guide to the IPC) may still be displayed in the scheme.

1/00	<b>Ammonia; Compounds thereof</b> ({ <a href="#">C01C 3/08</a> , <a href="#">C01C 3/14</a> , <a href="#">C01C 3/16</a> , <a href="#">C01C 3/20</a> take precedence})	1/0458	. . . . {Separation of NH <sub>3</sub> (during purge gas treatment <a href="#">C01C 1/0476</a> )}
	<b>NOTE</b>	1/0464	. . . . {by absorption in liquids, e.g. water}
	Complex ammine salts, e.g. [Pd(NH <sub>3</sub> ) <sub>4</sub> ]Cl <sub>2</sub> , are { also } classified in the relevant groups of subclasses <a href="#">C01D</a> - <a href="#">C01G</a> , according to the metal	1/047	. . . . {by condensation}
		1/0476	. . . . {Purge gas treatment, e.g. for removal of inert gases or recovery of H <sub>2</sub> }
1/003	. {Storage or handling of ammonia}	1/0482	. . . . {Process control; Start-up or cooling-down procedures}
1/006	. . {making use of solid ammonia storage materials, e.g. complex ammine salts}	1/0488	. . . . {Processes integrated with preparations of other compounds, e.g. methanol, urea or with processes for power generation}
1/02	. Preparation, {purification} or separation of ammonia	1/0494	. . . {using plasma or electric discharge}
1/022	. . {Preparation of aqueous ammonia solutions, i.e. ammonia water}	1/08	. . Preparation of ammonia from nitrogenous organic substances
1/024	. . {Purification}	1/083	. . . {from molasses (treatment of molasses in general <a href="#">C13B 50/006</a> )}
1/026	. . {Preparation of ammonia from inorganic compounds}	1/086	. . . {from urea}
1/028	. . . {from ammonium sulfate or sulfite}	1/10	. . Separation of ammonia from ammonia liquors, e.g. gas liquors {(as part of the ammonia synthesis process <a href="#">C01C 1/04</a> )}
1/04	. . Preparation of ammonia by synthesis {in the gas phase}(preparation or purification of gas mixtures for ammonia synthesis { <a href="#">C01B 3/025</a> })	1/12	. . Separation of ammonia from gases and vapours {(as part of the ammonia synthesis process <a href="#">C01C 1/04</a> )}
1/0405	. . . {from N <sub>2</sub> and H <sub>2</sub> in presence of a catalyst}	1/14	. . . Saturators
1/0411	. . . . {characterised by the catalyst}	1/16	. Halides of ammonium
1/0417	. . . . {characterised by the synthesis reactor, e.g. arrangement of catalyst beds and heat exchangers in the reactor (arrangement of several reactors <a href="#">C01C 1/0405</a> ; fixed-bed reactors in general <a href="#">B01J 8/02</a> )}	1/162	. . {Ammonium fluoride}
1/0423	. . . . . {Cold wall reactors}	1/164	. . {Ammonium chloride}
1/0429	. . . . . {Fluidized or moving bed reactors}	1/166	. . {Ammonium bromide}
1/0435	. . . . . {Horizontal reactors}	1/168	. . {Ammonium iodide}
1/0441	. . . . . {Reactors with the catalyst arranged in tubes}	1/18	. Nitrates of ammonium
1/0447	. . . . . {Apparatus other than synthesis reactors}	1/185	. . {Preparation}
1/0452	. . . . . {Heat exchangers}	1/20	. Sulfides; Polysulfides
		1/22	. Sulfites of ammonium
		1/24	. Sulfates of ammonium ( <a href="#">C01C 1/14</a> takes precedence)

- 1/242 . . Preparation from ammonia and sulfuric acid or sulfur trioxide
- 1/244 . . Preparation by double decomposition of ammonium salts with sulfates
- 1/245 . . Preparation from compounds containing nitrogen and sulfur
- 1/246 . . . from sulfur-containing ammonium compounds
- 1/247 . . . . by oxidation with free oxygen
- 1/248 . . Preventing coalescing or controlling form or size of the crystals
- 1/249 . . Deacidifying {or drying} the crystals
- 1/26 . Carbonates or bicarbonates of ammonium
- 1/28 . Methods of preparing ammonium salts in general
- NOTES**
1. This group does not cover ammonium salts of complex acids (other than complex cyanides) containing a metal in the anion, which are covered by the relevant groups of subclasses [C01D](#) - [C01G](#), according to the metal.
  2. Salts of polybasic acids with ammonium and a metal as cations are classified as though the ammonium were hydrogen.
- 3/00 Cyanogen; Compounds thereof**
- 3/001 . {Preparation by decomposing nitrogen-containing organic compounds, e.g. molasse waste or urea (by distillation of carbamates [C01C 3/02](#), [C01C 3/08](#), [C01C 3/14](#), [C01C 3/16](#); by decomposing formamide or ammonium formate [C01C 3/0204](#))}
- 3/002 . {Synthesis of metal cyanides or metal cyanamides from elementary nitrogen and carbides}
- 3/003 . {Cyanogen}
- 3/004 . {Halogenides of cyanogen}
- 3/005 . {Thiocyanogen}
- 3/006 . {Sulfur dicyanide}
- 3/007 . {Ammonium cyanide}
- 3/008 . {Cyanazide}
- 3/02 . Preparation, {separation or purification} of hydrogen cyanide {([C01C 3/001](#) takes precedence)}
- 3/0204 . . {from formamide or from ammonium formate}
- 3/0208 . . {Preparation in gaseous phase}
- 3/0212 . . . {from hydrocarbons and ammonia in the presence of oxygen, e.g. the Andrussov-process}
- 3/0216 . . . . {characterised by the catalyst used}
- 3/022 . . . . {Apparatus therefor}
- 3/0225 . . . . . {characterised by the synthesis reactor}
- 3/0229 . . . {from hydrocarbons and ammonia in the absence of oxygen, e.g. HMA-process}
- 3/0233 . . . . {making use of fluidised beds, e.g. the Shawinigan-process}
- 3/0237 . . . {from carbon monoxide and ammonia}
- 3/0241 . . . {from alcohols or aldehydes}
- 3/0245 . . . {from organic nitriles, e.g. acetonitrile}
- 3/025 . . . {by using a plasma}
- 3/0254 . . {from cyanates or from thiocyanates}
- 3/0258 . . {from cyanamides or derivatives thereof}
- 3/0262 . . {from cyanides}
- 3/0266 . . . {from simple alkali or alkaline earth metal cyanides}
- 3/027 . . . . {Alkali metal cyanides}
- 3/0275 . . . . {Alkaline earth metal cyanides}
- 3/0279 . . . {from ammonium cyanide}
- 3/0283 . . . {from simple or complex cyanides of the noble metals}
- 3/0287 . . . {from simple or complex cyanides of other transition metals, e.g. from iron cyanides}
- 3/0291 . . . {from simple or complex cyanides of other metals}
- 3/0295 . . {Purification}
- 3/04 . . Separation from gases
- 3/06 . Stabilisation of hydrogen cyanide
- 3/08 . Simple or complex cyanides of metals  
{([C01C 3/001](#), [C01C 3/002](#) take precedence)}
- 3/10 . . Simple alkali metal cyanides
- 3/11 . . Complex cyanides
- 3/12 . . Simple or complex iron cyanides
- 3/14 . Cyanic {or isocyanic} acid; Salts thereof  
{([C01C 3/001](#) takes precedence)}
- 3/145 . . {Isocyanic acid; Salts thereof}
- 3/16 . Cyanamide; Salts thereof {([C01C 3/001](#), [C01C 3/002](#) takes precedence); dicyandiamide [C07C 279/28](#)}
- 3/18 . . Calcium cyanamide
- 3/20 . Thiocyanic acid; Salts thereof {([C01C 3/001](#) takes precedence)}